Phone: +966 03-860-2854 Fax: +966 03-860-2879 E-mail: vohra@kfupm.edu.sa

FEB/ Vol 20/ No 5a/ 2011 – pages 1308 – 1313

# **THE INHIBITORY EFFECTS OF SOME PESTICIDES ON HUMAN ERYTHROCYTE GLUCOSE-6- PHOSPHATE DEHYDROGENASE ACTIVITY (***IN VITRO***)**

**Ersin Hopa, Selma Sinan and Yusuf Turan\***

Department of Biology, Faculty of Art and Science, Balıkesir University, 10145 Balıkesir, Turkey

# **ABSTRACT**

Human erythrocyte glucose-6-phosphate dehydrogenase was purified to apparent homogeneity by salting out with ammonium sulfate and applying one chromatographic step with the commercially available resin 2',5'-ADP Sepharose 4B. The enzyme, having a specific activity of 70.7 U/mg protein, was purified 7069 fold with an overall enzyme yield of 33.6%. The purity and molecular mass of the protein was confirmed with SDS-PAGE. The purified glucose-6-phosphate dehydrogenase was effectively active on glucose-6-phosphate and NADP+ with *K*m values of 0.22 and 0.14 mM and *V*max values of 1.94 and 2.76 U/mg, respectively. The *in vitro* effects of commonly used pesticides Glyphosate<sup>TM</sup> (N-(Phosphonomethyl) glycine) and  $2,4$ -D<sup>TM</sup> (2,4-Dichlorophenoxyacetic acid) were determined on the purified glucose-6-phosphate dehydrogenase. Both pesticides were effective inhibitors on the activity with  $IC_{50}$ values of 32.35 and 38.34 mM, respectively. The interaction kinetics of Glyphosate<sup>TM</sup> and  $2,4-D^{TM}$  with the purified enzyme indicated uncompetitive and competitive inhibition patterns with *K*i values of 13.45 and 8.45 mM, respectively.

**KEYWORDS:** Glucose-6-phosphate dehydrogenase, purification, pesticide, inhibition.

# **1. INTRODUCTION**

Glucose-6-phosphate dehydrogenase (EC 1.1.1.49; G6PD) is the first and key enzyme of the pentose phosphate metabolic pathway, catalysing the conversion of glucose-6phosphate to 6-phosphogluconate in the presence of NADP<sup>+</sup>. The fundamental physiological function of G6PD is the production of NADPH and ribose-5-phosphate, which are essential for reductive biosynthetic metabolisms and

nucleic acid synthesis, respectively [1,2]. The major role of NADPH in erythrocytes is the regeneration of reduced

#### \* Corresponding author

glutathione, which prevents hemoglobin denaturation, preserves the integrity of red blood cells' membrane sulfydryl groups, and detoxifies hydrogen peroxide and oxygen radicals in and on the red blood cells. Moreover, NADPH is also functional in cell membrane protection and cell de-toxification from xenobiotics, through the glutathione re-ductase-peroxidase system and mixedfunction oxidases [3-6].

The amount and variety of pesticides used have increased tremendously in recent years. This increase has caused a positive effect on crop production, however, certain pesticides, their residues, metabolites and/or contaminants have created many unforeseen adverse effects on the environment. Under some conditions, pesticides may be present in very low concentrations which have no immediate detectable effect. These small amounts of chemicals can cause sublethal (chronic) damage to organisms and this is more insidious and difficult to define than acute toxicity. Sublethal effects may be further enhanced by persistent pesticides which are accumulated in the organisms and magnified in the food chain [7-10]. Thus, the aim of this present study was mainly to purify G6PD from human erythrocytes and to investigate the *in vitro* inhibitory effects of more commonly used pesticides on this purified enzyme.

# **2. MATERIALS AND METHODS**

#### **2.1. Chemicals**

Glucose-6-phosphate, NADP<sup>+</sup>, NADPH, protein assay reagents and chemicals for electrophoresis were obtained from Sigma Chem. 2',5'-ADP Sepharose 4B was purchased from Pharmacia. Glyphosate<sup>TM</sup> (N-(phosphono-



methyl) glycine) and  $2,4-D<sup>TM</sup>$  (2,4-Dichlorophenoxyacetic acid) were provided from authorized local pesticide shop. All other chemicals were of the best available grade.

#### **2.2. Preparation of the hemolysate**

Appropriate amount of fresh blood from human was collected in the tubes containing anticoagulant EDTA and sample tubes were centrifuged at 3000g for 10 min. The plasma and leukocyte coat were removed by pipette. The package of erythrocytes was washed three times with isotonic KCl solution, then hemolysed with 5 volumes of icecold distilled water containing 2.5 mM EDTA and 0.7 mM β-ME. The hemolysate was centrifuged at 15000 g for 20 min at 4 $\rm{^0C}$  to remove the ghost and intact cells [11].

#### **2.3. Ammonium sulfate fractionation and dialysis**

The hemolysate was treated with solid ammonium sulfate to obtain the 30-70% fraction after centrifuging at 15000g for 30 min. The precipitate was dissolved in 50 mM potassium phosphate buffer, pH 7.0, and then dialised at 4  $\rm{^0C}$  in 50 mM potassium acetate/50 mM potassium phosphate buffer, pH 7.0 [12].

### **2.4. 2',5'-ADP Sepharose 4B affinity chromatography**

All purification steps were performed at 4°C unless otherwise stated. Two grams of dry 2',5'-ADP Sepharose 4B for 10 ml of bed volume was several times washed, to remove impurities and conditioning the gel, in 300 ml of distilled water. Following removal of the foreign bodies and air bubbles in the gel, it was resuspended in 0.1 M potassium acetate + 0.1 M potassium phosphate buffer, pH 6.0, at a ratio of 3 gel to1 buffer, and packed in a column  $(1x10 \text{ cm})$ . After setting of the gel, the column was pre-equilibrated with the same buffer prior to loading the enzyme solution. The dialised enzyme solution was loaded on the column and then was sequentially washed with 25 ml of 0.1 M potassium acetate  $+$  0.1 M potassium phosphate buffer, pH 6.0, and 25 ml of 0.1 M potassium acetate  $+ 0.1$  M potassium phosphate buffer, pH 7.85. The washing with 0.1 M potassium chloride + 0.1 M potassium phosphate buffer, pH 7.85 was continued until the absorbance becomes almost zero at 280 nm. The enzyme was eluted with a solution of 80 mM potassium phosphate  $+ 80$  mM potassium chloride  $+ 0.5$  mM NADP<sup>+</sup> + 10 mM EDTA (pH 7.85) at a flow rate of 20 ml/h, and 1 ml fractions were collected. The proteins containing the highest glucose-6-phosphate dehydrogenase activity were combined and used as purified enzyme for subsequent studies after confirming homogeneity by gel electrophoresis [2, 12].

#### **2.5. Glucose-6-phosphate dehydrogenase assay and protein determination**

The enzyme activity was routinely measured by Beutler's method [11]. One enzyme unit was defined as the enzyme amount that reduces  $1\mu$ mole NADP<sup>+</sup> per min under the assay conditions.

Quantitative protein determination was performed [13] using bovine serum albumin (BSA) as a standard.

#### **2.6. SDS Polyacrylamide gel electrophoresis (SDS-PAGE):**

Protein samples were fractionated on 12% SDS-PAGE gels [14] using a Minigel system (Bio-Rad Laboratories, USA). Gels were fixed, stained with Coomassie brilliant blue R-250 (Sigma), and destained using standard methods to detect protein bands.

#### **2.7. Determination of kinetic parameters and i***n vitro* **inhibition studies**

Various final concentrations of G6P were used to estimate the kinetic parameters *K*m and *V*max. Inhibition experiments were performed using G6P as substrate and different final concentrations of commonly used pesticides Glyphosate<sup>TM</sup> and  $2,4-D^{TM}$  as inhibitors. The double reciprocal Lineweaver–Burk plot was used to calculate the parameters. Activity % values of glucose-6-phosphate dehydrogenase for five different concentrations of each possible inhibitor were determined by regression analysis. Glucose-6-phosphate dehydrogenase activity without a possible inhibitor was accepted as 100% activity. The inhibitor concentration reduces the enzymatic activity by  $50\%$  (IC<sub>50</sub>) values) were determined from the graphs.

# **3. RESULTS AND DISCUSSION**

Human erythrocyte glucose-6-phosphate dehydrogenase was purified to apparent homogeneity by salting out with ammonium sulfate and subsequentially using 2',5'-ADP Sepharose 4B affinity chromatography. Upon fractionation of the G6PD active fractions with ammonium sulfate, 68% of the activity was obtained in the fraction saturated with 30-70% ammonium sulfate. This step removed the greater part of the contaminants and considerably decreased total protein amount. The precipitate with G6PD activity was dissolved and applied on 2',5'-ADP Sepharose 4B affinity chromatography column. Figure 1 shows the typical elution pattern of the enzyme activity on this affinity column. The enzyme activity and total protein concentrations were determined from all fractions collected. The fractions with the highest G6PD activity and the relatively lower protein contents were pooled. In order to reduce the number of the purification steps, this affinity chromatography purified the enzyme from remaining contaminants to apparent homogeneity, retaining 50% of the activity from the previous step. The G6PD was purified 7069 fold with an overall enzyme yield of 33.6% and a specific activity of 70.7 U/mg (Table 1).

SDS-PAGE analysis of the purified enzyme showed the presence of a single band with an apparent molecular mass of ca. 59 kDa, when stained with Coomassie brilliant blue (Figure 2).

The reaction kinetics of the purified G6PD were determined from Lineweaver–Burk plots using glucose-6 phosphate and NADP<sup>+</sup> as substrates with *K*m values of



0.22 and 0.14 mM and *V*max values of 1.94 and 2.76 U/mg, respectively. (Table 2). Affinity of the enzyme for  $NADP<sup>+</sup>$ was considerably higher than for glucose-6-phosphate. The higher G6PD affinity for  $NADP<sup>+</sup>$  have also been reported [15]. The inhibition kinetic experiments of the enzyme were

performed using glucose-6-phosphate as substrate, while commonly used pesticides Glyphosate<sup>TM</sup> and 2,4-D<sup>TM</sup> as inhibitors. Table 3 shows that the enzyme was uncompetitively and competitively inhibited by Glyphosate<sup>TM</sup> and



**FIGURE 1 - Purification of human erythrocyte glucose-6-phosphate dehydrogenase by affinity chromatography. The enzyme activity and total protein concentrations were determined, from all fractions collected, as described in Methods section. The enzyme activity was expressed as the enzyme amount that reduces 1µmole NADP+ per min under the assay conditions.**







**FIGURE 2 - SDS-PAGE of purified human erythrocyte glucose-6 phosphate dehydrogenase. The enzyme was electrophoresed at pH 8.3 on a 12% polyacrylamide gel and stained with Coomassie brilliant blue R-250. Lanes: 1, molecular weight standards (***β***galactosidase,116 kDa; bovine serum albumin, 66.2 kDa; egg albumin, 45 kDa; lactate dehyrogenase, 35 kDa; Rease Bsp981 (***E.coli***), 25 kDa;** *β***-lactoglobulin, 18.4 kDa; Lysozyme, 14.4 kDa; 2, purified human erythrocyte glucose-6-phosphate dehydrogenase).**

**TABLE 2 - Kinetic parameters of human erythrocyte glucose-6 phosphate dehydrogenase**



**TABLE 3 - Inhibitory effects of Glyphosate TM and 2,4-D TM on human erythrocyte glucose-6-phosphate dehydrogenase**



 $2.4 \text{-}D^{TM}$ , respectively. Glyphosate<sup>TM</sup> which is a widely used herbicide in agricultural fields, was the most effective inhibitor of the enzymatic activity with *K*i value of 13.45 mM (Figure 3) and  $IC_{50}$  of 32.35 mM (Figure 4). As shown in



Figures 5 and 6 the obtained *K*i and  $IC_{50}$  values of 2,4-D<sup>TM</sup> were 8.45 and 38.34 mM, respectively.

The essential role of G6PD activity in overall metabolism of organisms has been well known for several years. Glutathione is used by antioxidant defense mechanisms and its production depends on NADPH to be synthe-

sized in the pentose phosphate metabolic pathway mainly with G6PD catalytic activity. Thus, G6PD has been considered as an antioxidant enzyme [16]. Many chemicals at relatively low dosages affect the metabolism of biota by altering normal enzyme activity. In some of these interactions there is high reactivity involving a high degree effect



**FIGURE 3 - Lineweaver-Burk graph of G6PD in different glucose-6-phosphate and GlyphosateTM concentrations.**



**FIGURE 4 - Activity % curve of G6PD in different GlyphosateTM concentrations.**





**FIGURE 5 - Lineweaver-Burk graph of G6PD in different glucose-6-phosphate and 2,4-DTM concentrations.**



**FIGURE 6 - Activity % curve of G6PD in different 2,4-DTM concentrations.**

on the whole animal or plant. On the other hand, many chemicals affect the activity of many enzymes only to a moderate degree and it is presumed that the ultimate debilitating effect on the whole organism develops from a variety of nonspecific biochemical functions [17]. Therefore, Glyphosate<sup>TM</sup> and  $2.4-D^{TM}$  were chosen to investigate their effects on the G6PD activity. Glyphosate<sup>TM</sup> has been reported as the most used herbicide in the USA with 5-8 million pounds [2.5 to 4 kilotonnes] every year on lawns and yards and 85–90 million pounds annually in US agriculture [18]. 2,4- $D^{TM}$  is also the most widely used herbicide in the world, and the third most commonly used in North America [19]. Laboratory toxicology studies suggest that other ingredients combined with Glyphosate  $TM$  may have greater toxicity than Glyphosate  $TM$  alone. For example, a study comparing Glyphosate<sup>TM</sup> and Roundup<sup>TM</sup> which is the brand name of a systemic herbicide and contains the active ingredient  $Glyphosate^{TM}$ , found that Roundup<sup>TM</sup> had a greater effect on aromatase than Glyphosate<sup>TM</sup> alone [20]. Another study has shown that  $Roundup<sup>TM</sup>$  formulations and metabolic products cause the death of human embryonic, placental, and umbilical cells *in vitro* even at low concentrations. The effects are not proportional to Glyphosate<sup>TM</sup> concentrations but dependent on the nature of the adjuvants used in the formulation [21]. Also it has been reported that  $2.4 \text{-} D^{TM}$  causes cancer in humans and it's exposure substantially increased the risk of Non-Hodgkin's lymphoma and amyotrophic lateral sclerosis [22-24]. Since the strong inhibitory concentrations of these pesticides have been found rather low, both have negative effects on the metabolism through G6PD activity.

# **REFERENCES**

[1] Kuo, W. Lin, J. and Tang, T. K. (2000) Human glucose-6 phosphate dehydrogenase (G6PD) gene transforms nih 3t3 cells and induces tumors in nude mice. Int. J. Cancer. 85, 857-864.

- [2] Şentürk, M, Ceyhun, S. B. Erdoğan, O. and Küfrevioğlu, İ. Ö. (2009) *In vitro* and *in vivo* effects of some pesticides on glucose-6phosphate dehydrogenase enzyme activity from rainbow trout (*Oncorhynchus mykiss*) erythrocytes. Pesticide Biochem. and Physiol. 95, 95-99.
- [3] Deutsch, J. (1983) Glucose-6-phosphate dehydrogenase, in: H. U. Bergmeyer, J. Bergmeyer (Eds), Methods of Enzymatic Analysis, Vol. 3, Verlagsgerellschaff, VCH, pp. 190-196.
- [4] Barroso, J. B. Peragón, J. Garcia-Salguero, L. De La Higuera, M. and Lupiáñez, J. A. (1999) Variations in the kinetic behaviour of the NADPH-production system in different tissues of the trout when fed on an amino acid based diet at different frequencies. Int. J. Biochem. Cell B. 31, 277-290.
- [5] Celik, I. and Demir H. (2005) Effects of acute trichloroacetic acid treatment on the in-vivo activity of glucose-6-phosphate dehydrogenase and carbonic anhydrase in various tissues of rats. Fresen. Environ. Bull., 14, 256-258.
- [6] Gülçin, İ., Beydemir, Ş., Çoban, T.A. and Ekinci, D. (2008) The inhibitory effect of dantrolene sodium and propofol on 6 phosphogluconate dehydrogenase from rat erythrocyte. Fresen. Environ. Bull., 17(9A), 1283-1287.
- [7] Christensen, G. (1972) Effect of metal cations and other chemicals upon the in vitro activity of two enzymes in the blood plasma of white sucker, *catostomus commersoni* (lacépède). Chem. Biol. Int., 4, 351-361.
- [8] Christensen, G. Olson, D. and Riedel, B. (1982) Chemicals effect on the activity of eight enzymes. A review and a Discussion Relevant to Environmental Monitoring. Environmental Research. 29, 247-255.
- [9] Tandon, R. S. and Dubey, A. (1983) Toxic effect of two organophosphorus pesticides on fructose-1,6-diphosphate aldolase activity of liver, brain and gills of the freshwater fish *Clarias batrashus*, Environ. Poll. (Series A), 31, 1-7.
- [10] Wiszkowski, H. Kulamowicz, I. and Malinowska, A. (1986) The effect of malathion on RNA polymerase activity of cell nuclei and transcription product in lymphocyte culture. Environ. Res. 41, 372-377.
- [11] Beutler, E. (1971) Red Cell Metabolism: A Manual of Biochemical Methods, pp. 92-94, Grune and Stratton New York.
- [12] Ninfali, P. Orsenigo, T. Barociani, L. and Rapa, S. (1990) Rapid purification of glucose-6-phosphate dehydrogenase from mammal's erythrocyte. Prep. Biochem. 20, 297-309.
- [13] Lowry, O. H. Rosebrough, N. J. Farr A.L. and Randall, R. J. (1951) Protein measurement with the Folin-Phenol reagents. J. Biol. Chem. 193, 265-275.
- [14] Laemmli, U. K. (1970) Cleavage of structural proteins during the assembly of the head of bacteriophage T4. Nature. 227, 680-685.
- [15] Özmen, İ. Çiftçi, M. Küfrevioğlu, Ö.İ. and Çürük, M.A. (2004) Investigation of Glucose 6-Phosphate Dehydrogenase (G6PD) Kinetics for Normal and G6PD Deficient Persons and the Effects of some Drugs. J. Enzyme Inhib. and Med. Chem.19 (1), 45-50.
- [16] Reiter, R. J. Tang, L. Garcia, J. J. and Munoz-Hoyos, A. (1997) Pharmacological actions of melatonin in oxygen radical pathophysiology. Life Sci. 60, 2255-2271.
- [17] Çelik, I. Çamas, H. Arslan, O. Yegin, E. and Küfrevioglu, Ö. I. (1996) The effects of some pesticides on the activity of liv-

er and erythrocyte enzymes (*in vitro*). J. Env. Sci. Health. A31 (7), 1645-1649.

- [18] United States Environmental Protection Agency 2000–2001 Pesticide Market Estimates. http://www.epa.gov/pesticides/ pestsales/01pestsales/usage2001\_2.htm
- [19] United States Environmental Protection Agency 2000–2001 Pesticide Market Estimates. http://www.epa.gov/pesticides /pestsales/01pestsales/usage2001\_3.htm and http://www.24d.org
- [20] Richard S, Moslemi, S. Sipahutar, H. Benachour, N. and Seralini, G. E. (2005). Differential effects of glyphosate and roundup on human placental cells and aromatase. Environ. Health Perspect. 113 (6), 716–20.
- [21] Benachour, N. and Séralini, G. E. (2008). Glyphosate Formulations Induce Apoptosis and Necrosis in Human Umbilical, Embryonic and Placental Cells. Chem. Res. Toxicol. 22 (1), 97-105.
- [22] Ibrahim, M. A. Bond, G. G. Burke, T. Cole, A. P. Dost, F. N. Enterline, P. Gough, E. M. Greenberg, R. S. Halperin, W. E. McConnell, E. Munro, I. C. Swenberg, J. A. Zahm, S. H. and Graham, J. D. (1991) Weight of the evidence on the human carcinogenicity of 2,4-D. Environ. Health Perspect., 96, 213– 222.
- [23] Zahm, S. H. Weisenburger, D. D. Babbitt, P. A. Saal, R. C. Vaught, J. B. Cantor, K. P. and Blair, A. (1990) A Case-Control Study of Non-Hodgkin's Lymphoma and the Herbicide 2,4-Dichlorophenoxyacetic Acid (2, 4-D) in Eastern Nebraska. Epidemiology. 1 (5), 349-356.
- [24] Burns, C. J. Beardb, K. K. and Cartmill, J. B. (2001) Mortality in chemical workers potentially exposed to 2,4-dichlorophenoxyacetic acid (2,4-D) 1945-94: an update, Occup. Environ. Med. 58, 24-30.